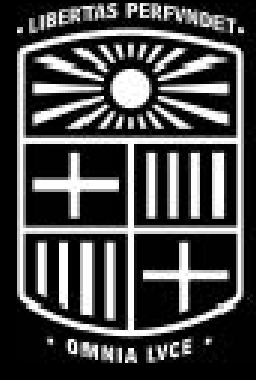


Exploring the Photoactive Properties of MXenes for Water Splitting

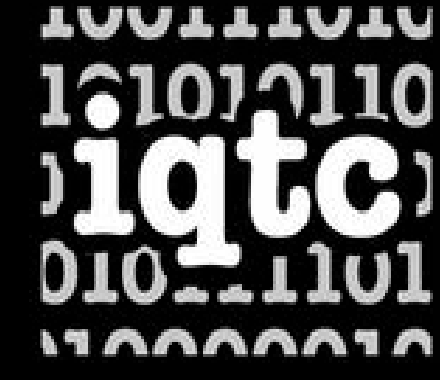


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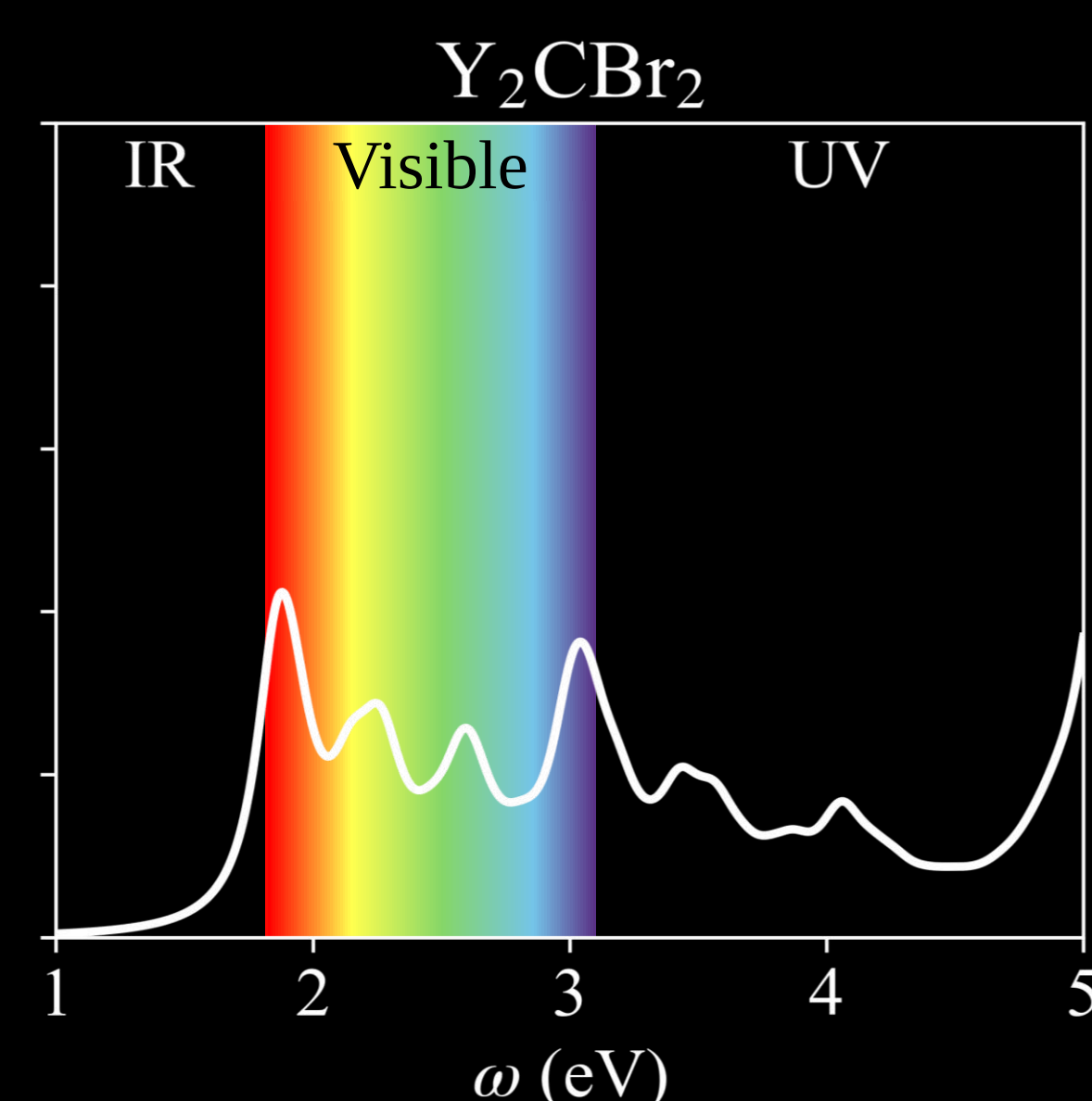
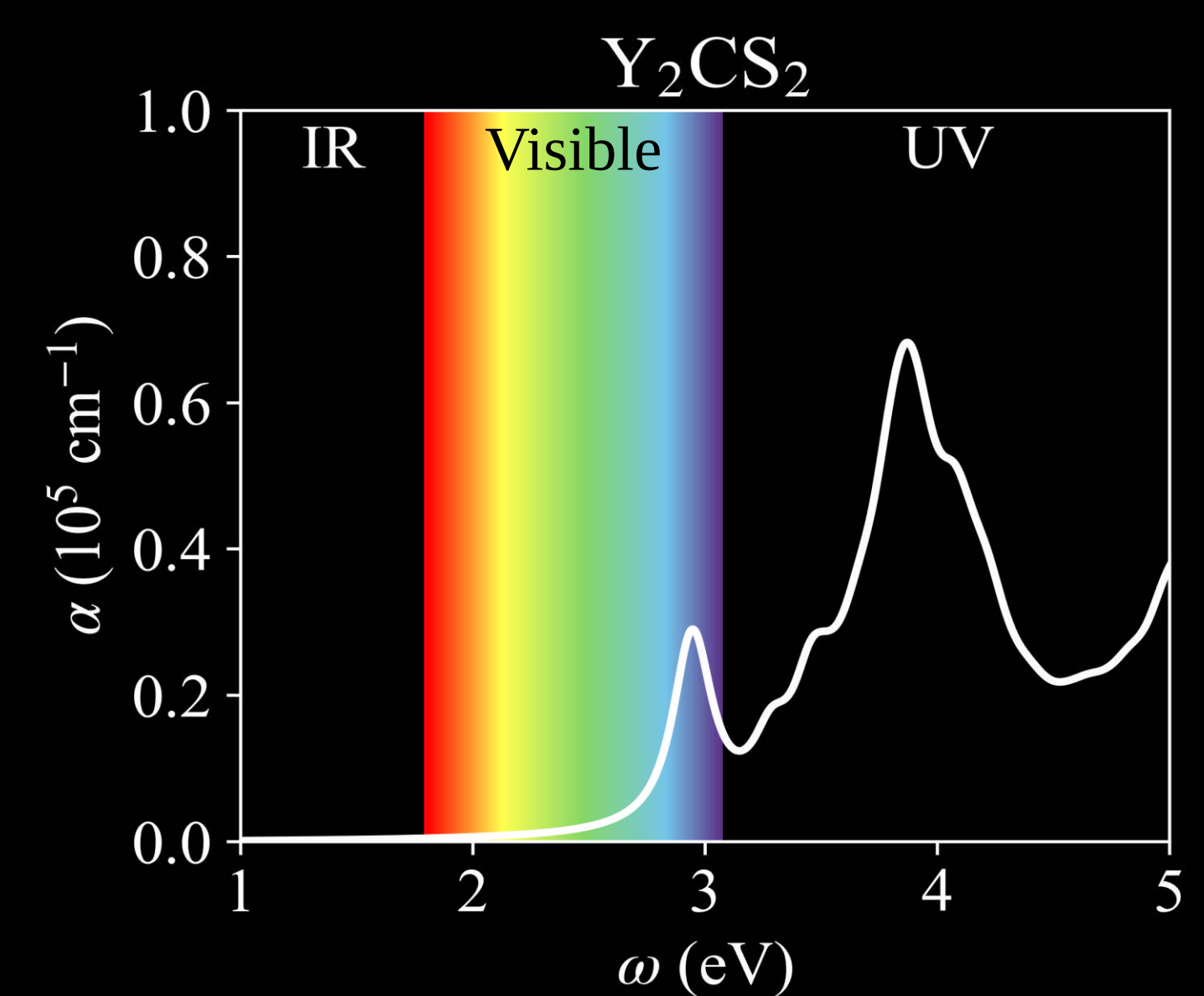
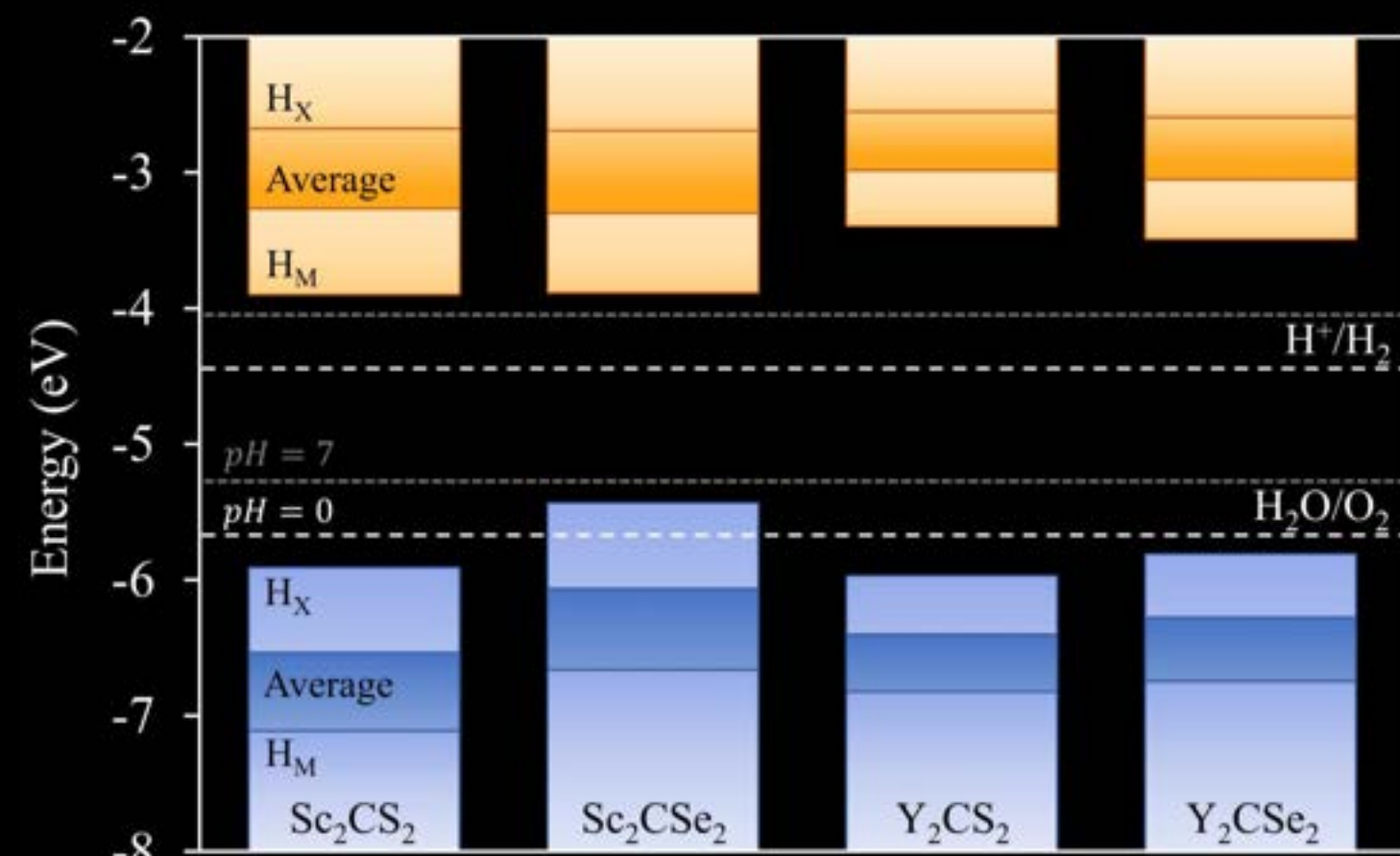
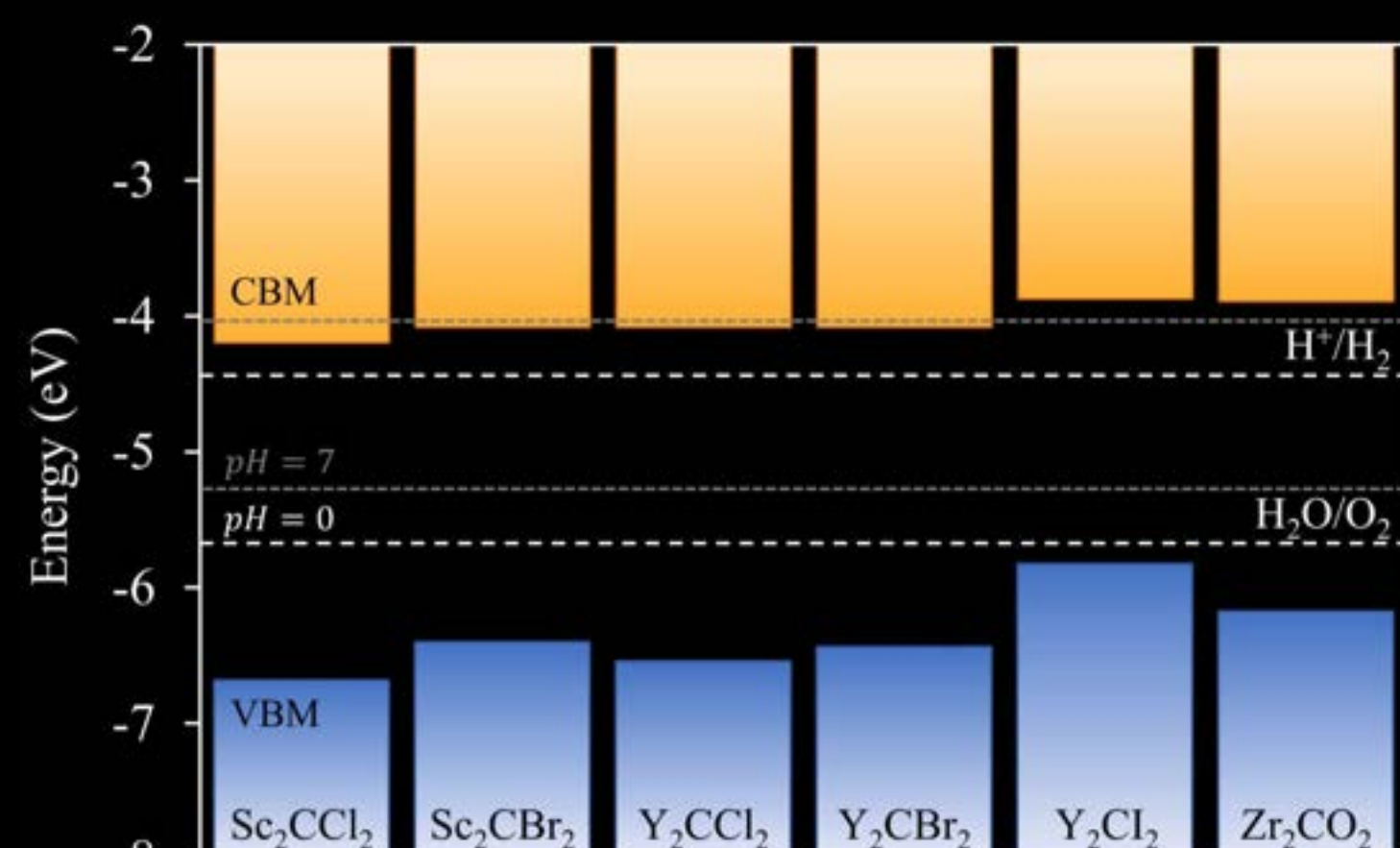
General Properties

- An effective photocatalyst needs good band alignment, efficient charge separation, minimal VBM/CBM overlap, anisotropic carrier mobilities, strong visible light absorption, and high solar-to-hydrogen (STH) efficiency.
- Cohesive energy, phonons → energetical and dynamical stability.

MXene	E_{coh} (eV/at.)	$E_{\text{g}}^{\text{PBE0}}$ (eV)	E_{opt} (eV)	Overlap (%)	η_{STH} (%)
Zr ₂ CO ₂	-7.68	2.26	2.87	27.9	2.5
Sc ₂ CS ₂	-5.52	3.23	3.34	35.2	0.9
Y ₂ CS ₂	-5.52	3.42	2.93	38.5	2.7
Sc ₂ CSe ₂	-5.16	2.75	3.15	31.9	1.4
Y ₂ CSe ₂	-5.17	3.21	2.99	34.5	2.3
Sc ₂ CCl ₂	-5.41	2.48	2.30	26.2	11.0
Y ₂ CCl ₂	-5.42	2.44	1.89	31.7	21.3
Sc ₂ CBr ₂	-5.09	2.31	2.31	27.1	10.9
Y ₂ CBr ₂	-5.13	2.36	1.85	30.7	22.6
Y ₂ Cl ₂	-4.77	1.94	1.79	25.2	12.2

Band Alignment

- $pH = 0$: ✓ band alignment for all cases, excepting Sc₂CSe₂ (on H_x surface).
- $pH = 7$: Some halide-terminated MXenes become unable to photocatalyze HER.
- H_{MX}: Janus → intrinsic $\vec{E} \rightarrow e^- - h^+$ separation.



CONCLUSIONS ✓

The photoactive potential of a group of 10 MXenes has been deeply explored using DFT calculations, showcasing robust stability, high charge carrier mobilities, strong visible light absorption, and promising solar-to-hydrogen efficiency. These features make them leading candidates for efficient water splitting photocatalysis, with H_M structures generally outperforming H_{MX} structures.

MXenes

$$(M_{n+1}X_nT_x)$$

M = Transition Metal (Groups 3 – 6)
X = C or N n = 1 – 4
T = Termination (Groups 16 – 17)

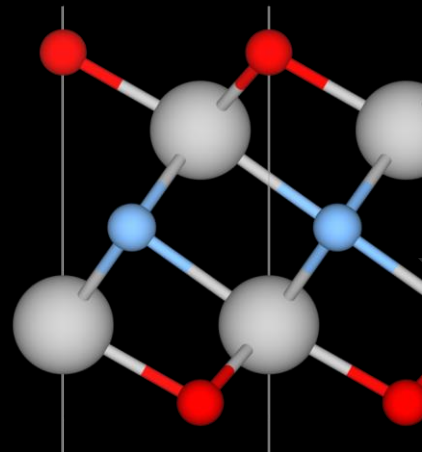
Promising candidates for
photocatalysing the water splitting
process and produce clean H₂.^[1,2]

Objective: Explore the photoactive effectiveness of MXenes in the water splitting process, through different photocatalytic properties.

10 promising
MXenes studied

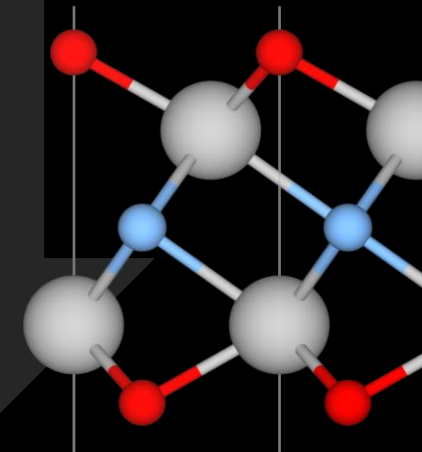
Sc₂CT₂, Y₂CT₂
(T = Cl, Br),
Y₂Cl₂, Zr₂CO₂

ABC H_M



Sc₂CT₂, Y₂CT₂
(T = S, Se)

ABC H_{MX}



Methods: DFT with PBE0 hybrid functional for electronic structure and GW-BSE for optical properties.

MXenes Approved

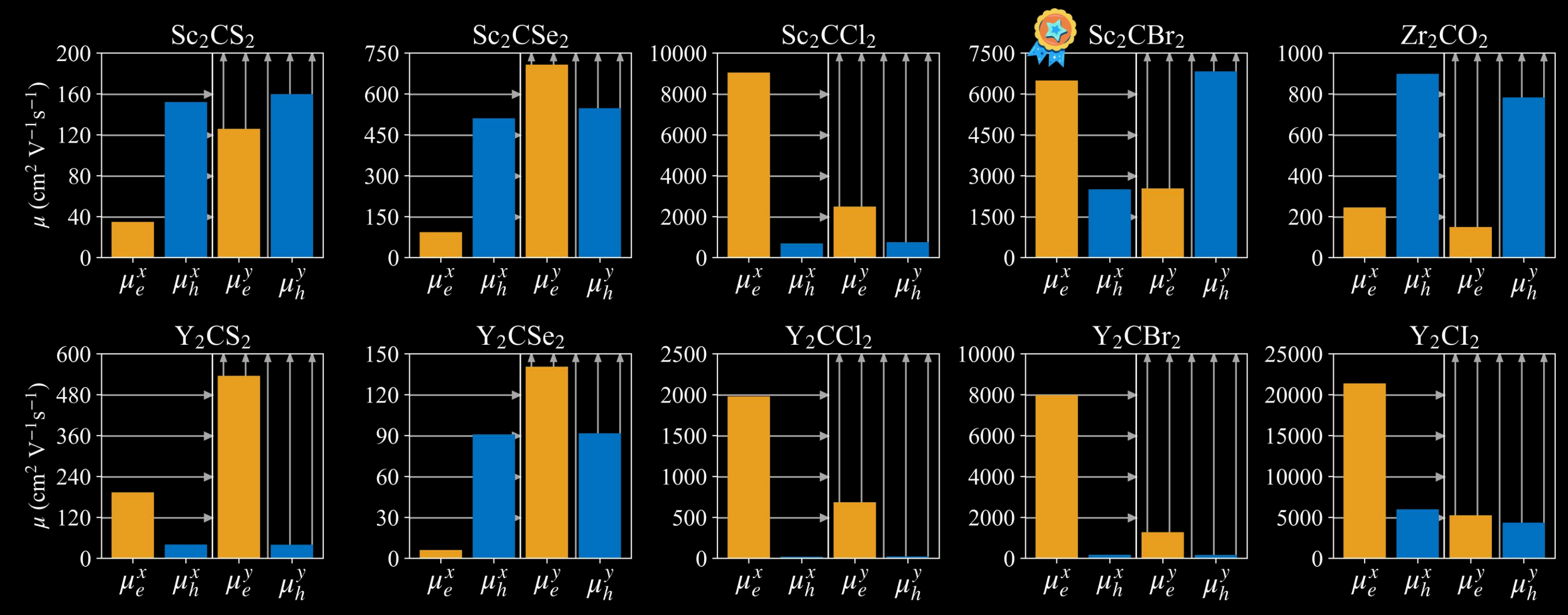


Optical Absorption

- H_M: Good optical absorption in visible range.
- H_{MX}: Optical absorption shifted to UV.
- $E_{\text{opt}}(\text{Y}) < E_{\text{opt}}(\text{Sc})$.
- BSE → Exciton → $E_b \approx 0.3 - 0.7$ eV.
- STH efficiency:
halide (11–23%) > chalcogen (1–3%).

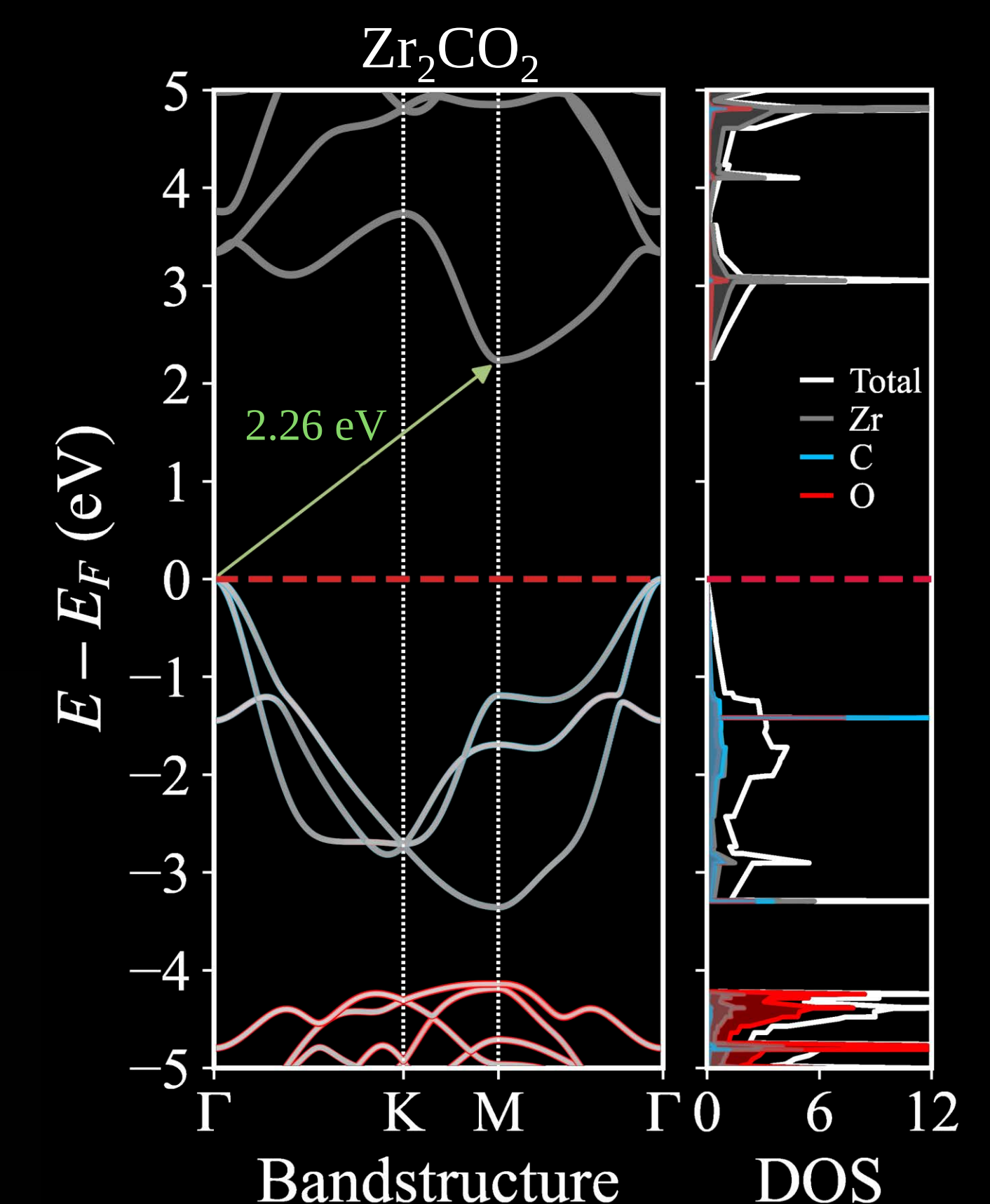
Charge Carrier Mobility

- Along x (zigzag $\wedge$) and y (armchair $\curvearrowright$).
- Anisotropic electron carrier mobility, $\mu_e^x \neq \mu_e^y$.
- H_{MX}: $\mu_e^x < \mu_e^y$, H_M: $\mu_e^x > \mu_e^y$.
- Isotropic hole carrier mobility, $\mu_h^x \approx \mu_h^y$.
- ⚠ Except for Sc₂CBr₂ → Asymmetric anisotropy both in e and h (charge separation ✓)



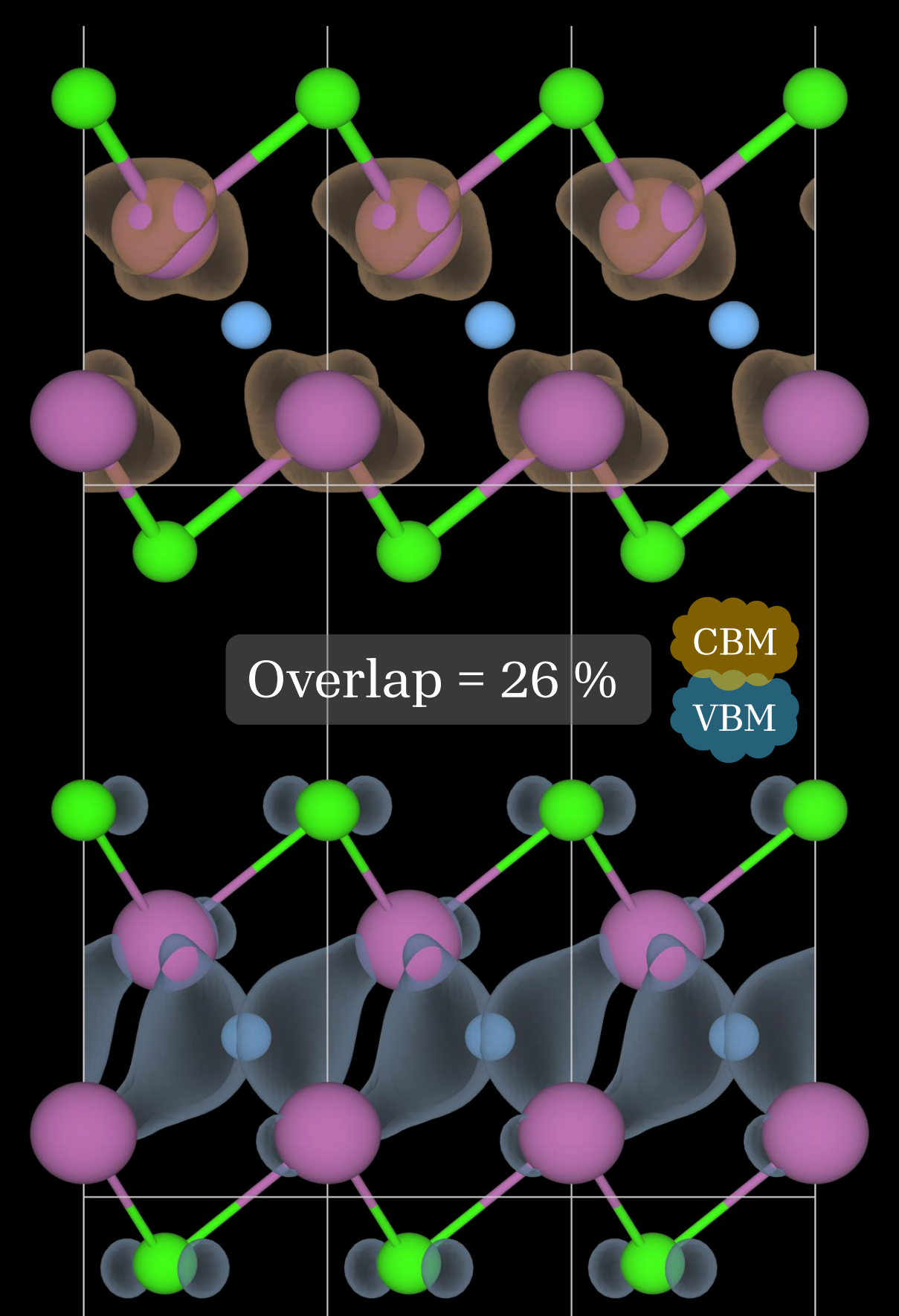
Electronic Structure

- Semiconductors $E_g \approx$ visible range.
- H_{MX} MXene structures: $E_g \approx$ UV.
- Indirect $\Gamma \rightarrow M$ (H_M) or $\Gamma \rightarrow K$ (H_{MX}) transitions.
- VB → C and M atoms, and T at lower energies.
- CB → d orbitals of M.



Charge Density

- Overlap(VBM, CBM) $\approx$ 25 – 38 %.
- H_M: VBM → $p(\text{C})$ orbitals, CBM → M layers.
- H_{MX}: VBM → H_x face, CBM → M-C layers.
↳ asymmetry in charge distribution.



REFERENCES

- [1] D. Ontiveros, F. Viñes, C. Sousa, *J. Mater. Chem. A*, 2023, 11, 13754–13764.
- [2] D. Ontiveros, S. Vela, F. Viñes, C. Sousa, *Energy Environ. Mater.*, 2024, 7, e12774.

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